

Feng Xu and Wei-Xiao Hu\*

College of Pharmaceutical Science, Zhejiang  
University of Technology, Hangzhou 310032,  
People's Republic of China

Correspondence e-mail: huyang@mail.hz.zj.cn

## Key indicators

Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.064  
 $wR$  factor = 0.167  
Data-to-parameter ratio = 16.1For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

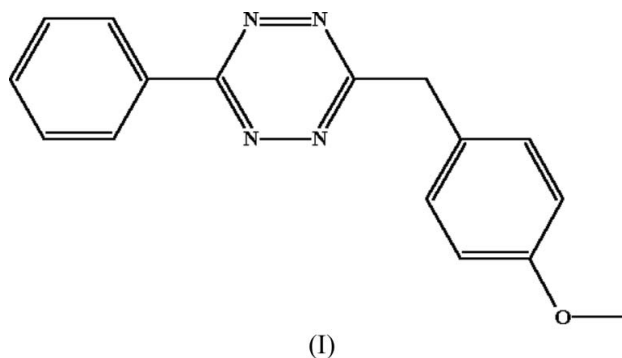
## 6-(4-Methoxyphenyl)-3-phenyl-1,2,4,5-tetrazine

In the title molecule,  $\text{C}_{16}\text{H}_{14}\text{N}_4\text{O}$ , the tetrazine ring makes dihedral angles of  $6.77$  (12) and  $70.5$  (5) $^\circ$  with the phenyl and methoxy-substituted benzene rings, respectively. The crystal packing exhibits  $\pi$ - $\pi$  stacking interactions and weak intermolecular  $\text{C}-\text{H}\cdots\text{N}$  hydrogen bonds.

Received 12 January 2007  
Accepted 19 January 2007

## Comment

1,2,4,5-Tetrazine derivatives have high potential for biological activity, possessing a wide spectrum of antiviral and antitumor properties. They have been widely used in pesticides and herbicides (Sauer, 1996). In our search for structure-activity relationships in 1,2,4,5-tetrazine derivatives (Hu *et al.*, 2004, 2005), we obtained red crystals of the title compound, (I).



In (I) (Fig. 1), the tetrazine ring (centroid  $Cg1$ ) makes dihedral angles of  $70.5$  (5) and  $6.77$  (12) $^\circ$  with the aromatic rings  $\text{C}8-\text{C}13$  (centroid  $Cg2$ ) and  $\text{C}14-\text{C}19$  (centroid  $Cg3$ ), respectively. The crystal packing exhibits weak intermolecular  $\text{C}-\text{H}\cdots\text{N}$  hydrogen-bonding (Table 1) and  $\pi$ - $\pi$  stacking interactions [indicated by the short distances  $Cg1\cdots Cg2^{\text{ii}} = 3.694$  (3) and  $Cg1\cdots Cg3^{\text{iii}} = 3.715$  (2) Å; symmetry codes: (ii)  $-x, -y, 1 - z$ ; (iii)  $1 - x, \frac{1}{2} - y, \frac{3}{2} - z$ ].

## Experimental

With sulfur (1.0 g) as catalyst, 85% hydrazine hydrate (10 ml, 170 mmol) was dropped into a mixed anhydrous ethanol (15 ml) solution of 4-methoxybenzyl cyanide (50 mmol) and benzonitrile (50 mmol) at 295 K. After refluxing for 3 h, the mixture was cooled to room temperature and the resulting solid product was filtered off. It was dissolved in diethyl ether (15 ml), and oxidized by sodium nitrate (14 mmol) and acetic acid (14 mmol) for 2 h to afford the product, which was purified by preparative thin-layer chromatography over silica-gel PF254 (2 mm) (cyclohexane-dichloromethane, 1:1) to give red single crystals. X-ray quality single crystals of (I) were obtained by slow evaporation at room temperature of a tetrahydrofuran-anhydrous ethanol (4:1 *v/v*) solution.

## Crystal data

$C_{16}H_{14}N_4O$   
 $M_r = 278.31$   
 Monoclinic,  $P2_1/c$   
 $a = 8.599$  (4) Å  
 $b = 15.255$  (6) Å  
 $c = 11.794$  (4) Å  
 $\beta = 112.67$  (2)°  
 $V = 1427.5$  (9) Å<sup>3</sup>

$Z = 4$   
 $D_x = 1.295$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
 Prism, red  
 $0.30 \times 0.25 \times 0.20$  mm

## Data collection

Bruker SMART CCD area-detector  
 diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan  
 (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.975$ ,  $T_{\max} = 0.983$

6907 measured reflections  
 3083 independent reflections  
 1923 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.062$   
 $\theta_{\text{max}} = 27.0^\circ$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.064$   
 $wR(F^2) = 0.167$   
 $S = 1.00$   
 3083 reflections  
 192 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0862P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.28$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.32$  e Å<sup>-3</sup>  
 Extinction correction: SHELXL97  
 Extinction coefficient: 0.160 (10)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C12-H12 \cdots N5^i$	0.93	2.56	3.413 (2)	152

Symmetry code: (i)  $x, -y + \frac{1}{2}, z - \frac{1}{2}$ .

Methyl H atoms were placed in calculated positions, with  $C-H = 0.96$  Å and  $U_{\text{iso}}(H) = 1.5U_{\text{eq}}(C)$ . Other H atoms were placed in calculated positions, with  $C-H = 0.93-0.97$  Å, and refined in riding mode, with  $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ .

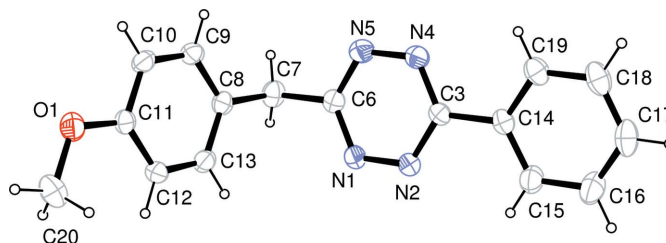


Figure 1

The molecular structure of (I), shown with 30% probability displacement ellipsoids (arbitrary spheres for H atoms).

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1999); software used to prepare material for publication: SHELXL97.

The authors are grateful to the National Natural and Scientific Foundation for financial support (grant No. 20272053).

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